

REACTIONS OF o-TOLYL ISOCYANIDE WITH ISOCYANATE AND ISOTHIOCYANATE

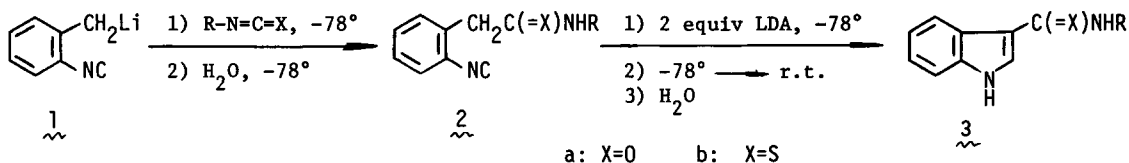
SYNTHESES OF N-SUBSTITUTED INDOLE-3-CARBOXAMIDES AND INDOLE-3-THIOCARBOXAMIDES

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Summary: o-Lithiomethylphenyl isocyanide is reacted at -78° with isocyanates and isothiocyanates to produce o-isocyanophenyl-acetamides (2a) and -acetothioamides (2b). Isocyanides 2a and 2b are cyclized to indole-3-carboxamides (3a) and -3-thiocarboxamides (3b) via lithiation, respectively. Isocyanides 2a are also cyclized by Cu_2O catalyst to produce 4,5-dihydro-1,3-benzodiazepin-4-ones (4a) with 3a.

In the preceding papers^{1,2} we reported that o-lithiomethylphenyl isocyanide (1), which is generated in situ at -78° from o-tolyl isocyanide and lithium diisopropylamide (LDA), was a versatile intermediate for the preparation of indoles and the related compounds. In this paper we wish to report an elaboration of o-lithiomethylphenyl isocyanide (1) to N-substituted indole-3-carboxamides (3a) and indole-3-thiocarboxamides (3b), in which 1 was reacted with isocyanate and isothiocyanate, followed by cyclization of the resulting N-substituted o-isocyanophenylacetamides (2a) and o-isocyanophenylacetothioamides (2b).



A sample procedure is illustrated by a synthesis of N-butyl indole-3-carboxamide. To a deep red solution of o-lithiomethylphenyl isocyanide in diglyme, which was prepared in situ at -78° by treating 176 mg (1.5 mmol) of o-tolyl isocyanide with LDA (3.0 mmol) in 4ml of diglyme according to the reported procedure,¹ was dropwise added 149 mg (1.5 mmol) of n-butyl isocyanate. The deep red color of 1 immediately turned to light red. After the reaction mixture was stirred for 30 min at -78° , it was quenched with aq NH_4Cl , extracted with ether, dried over anhydrous Na_2SO_4

and evaporated. The residue was distilled using Kugelrohr to afford N-butyl o-isocyanophenylacetamide (2a-i) in 70% yield (bp 125°/0.4 mmHg ; mp 95°), 2a-i : IR (neat) 3300, 2125, 1637 cm^{-1} ; NMR (CDCl_3 with TMS) δ 0.87 (t, 3H), 1.1~1.5 (m, 4H), 3.1 (m, 3H), 3.43 (s, 2H), 7.1~7.3 (m, 4H).

Next, 216 mg (1.0 mmol) of N-butyl o-isocyanophenylacetamide (2a-i) in 0.5 ml of diglyme was dropwise added to a solution of LDA (2 mmol) in 2.5 ml of diglyme at -78° with stirring. After the mixture was stirred for 30 min at -78° and then allowed to warm up to room temperature, it was poured into aq NH_4Cl and extracted with ether. The ether solution was concentrated in vacuo, and the residue was chromatographed on silica gel with 4:1 CHCl_3 -AcOEt to give N-butyl indole-3-carboxamide (3a-i) (Tlc Rf=0.69) in 55% yield [mp 135° (lit³ mp 131~131.5)] (Table 1).⁴

N-substituted indole-3-thiocarboxamides (3b) were prepared according to the same procedure except for that isothiocyanate was used in place of isocyanate. N-Substituted indole-3-thiocarboxamides can be also prepared in moderate yield in one-pot without isolating N-substituted o-isocyanophenylacetothioamides (2b) by successive addition of isothiocyanate and 2 equiv of LDA to o-lithiomethylphenyl isocyanide at -78°. However, the attempts to prepare N-substituted

Table 1. Preparations of N-Substituted o-Isocyanophenylacetamide (2a) and o-Isocyanophenylacetothioamides (2b)

R-N=C=X	Product (%) ^a
	<u>2a</u> or <u>2b</u>
n-C ₄ H ₉ -N=C=O	<u>2a-i</u> (70)
c-C ₆ H ₁₁ -N=C=O	<u>2a-ii</u> (55)
tert-C ₄ H ₉ -N=C=O	<u>2a-iii</u> (53)
Ph-N=C=O	<u>2a-iv</u> (50)
c-C ₆ H ₁₁ -N=C=S	<u>2b-i</u> (96)
Ph-N=C=S	<u>2b-ii</u> (82)

a) Isolated yields

indole-3-carboxamides (3a) in one-pot were unsuccessful. Some syntheses of N-substituted indole-3-carboxamides (3a) and indole-3-thiocarboxamides (3b) were summarized in Table 2.⁵

N-Substituted o-isocyanophenylacetamides (2a) were also cyclized by Cu_2O catalyst to produce N-substituted indole-3-carboxamides (3a) and N-substituted 4,5-dihydro-1,3-benzodiazepin-4-ones (4a),⁶ the

ratio of 3a and 4a depending upon substituent on the amide nitrogen.

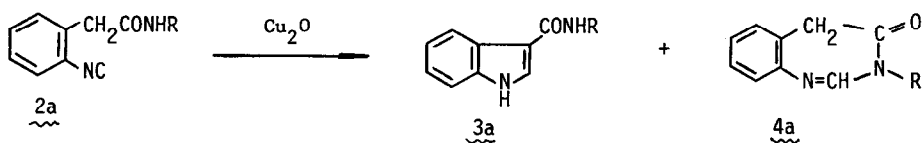
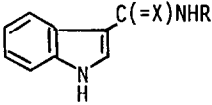


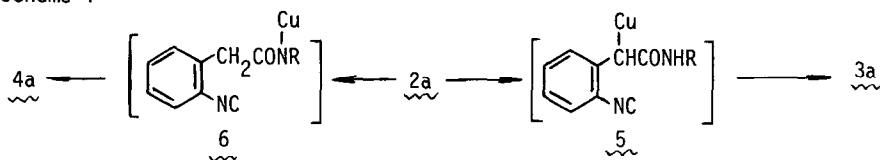
Table 2. Preparations of N-Substituted Indole-3-carboxamides (3a) and Indole-3-thiocarboxamides (3b)

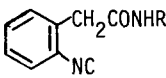
			Yield (%) ^a
R=n-C ₄ H ₉ ,	X=O	(<u>3a-i</u>)	55
R=c-C ₆ H ₁₁ ,	X=O	(<u>3a-ii</u>)	60
R=tert-C ₄ H ₉ ,	X=O	(<u>3a-iii</u>)	57
R=Ph,	X=O	(<u>3a-iv</u>)	68
R=c-C ₆ H ₁₁ ,	X=S	(<u>3b-i</u>)	60 (52) ^b
R=Ph,	X=S	(<u>3b-ii</u>)	58
R=CH ₂ =CHCH ₂ -,	X=S	(<u>3b-iii</u>)	— (58) ^b

a) Isolated yields from 2a or 2b

b) Overall yields from o-tolyl isocyanide (one-pot reaction)

Scheme 1

Table 3. Cu₂O Catalyzed Cyclizations ofN-Substituted o-Isocyanophenylacetamides (2a)

		Reaction Time (hr)	Products (%) ^a	
			<u>3a</u>	<u>4a</u>
R=n-C ₄ H ₉	(<u>2a-i</u>)	10 ^b	0	85
R=c-C ₆ H ₁₁	(<u>2a-ii</u>)	60 ^c	25	58
R=tert-C ₄ H ₉	(<u>2a-iii</u>)	60 ^c	20 ^d	0
R=Ph	(<u>2a-iv</u>)	10 ^b	75	0

a) Isolated yields. b) 0.2 equiv of Cu₂O was used. c) 1 equiv of Cu₂O was used.d) 60% of the starting amide (2a-iii) was recovered.

Some results on the Cu₂O-induced cyclization of N-substituted o-isocyanophenylacetamides (Table 3) seem to indicate that the less bulky alkyl substituents on the amide nitrogen favor the formation of 4a.

The formations of 3a and 4a in the Cu₂O-induced cyclizations of 2a may be explained in terms of intramolecular isonitrile insertion^{7,8} into the carbon-copper and the nitrogen-copper linkages of the organocopper intermediates 5 and 6, respectively, as shown in Scheme 1.

The Cu_2O induced cyclization of 2a was carried out as follows. A mixture of 242 mg (1 mmol) of N-cyclohexyl o-isocyanophenylacetamide (2a-ij), 143 mg (1 mmol) of Cu_2O and 10 ml of benzene was heated at reflux for 60 hr under nitrogen. The reaction mixture was filtered and the filtrate was distilled using Kugelrohr to afford N-cyclohexyl 4,5-dihydro-1,3-benzodiazepin-4-one (4a-ij) in 58% yield (bp $160^\circ/0.6$ mmHg; mp 115°). The residue was chromatographed on silica gel with 1:1 CHCl_3 -AcOEt to give N-cyclohexyl indole-3-carboxamide (3a-ij) (Tlc Rf=0.43) in 25% yield. 4a-ij: IR (KBr disk) 1666, 1622 cm^{-1} ; NMR (CDCl_3 with TMS) δ 1.1~1.8 (m, 10H), 3.37 (s, 2H), 4.3 (m, 1H), 7.10 (m, 4H), 7.45 (s, 1H).

Cyclization of 2b with Cu_2O catalyst failed.

References and Notes

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2. Y. Ito, K. Kobayashi and T. Saegusa, Tetrahedron Lett., 1978, 2087.
3. J. T. Carlock, J. S. Bradshaw, B. Stanovnik and M. Tišler, J. Org. Chem., 42, 1883 (1977).
4. 2b-i was recrystallized from 1:1 n-hexane and ether. mp $70\sim 72^\circ$. IR (KBr disk) 3260, 2120, 1120 cm^{-1} ; NMR (CDCl_3 with TMS) δ 1.1~1.9 (m, 10H), 3.91 (s, 2H), 4.2 (m, 1H), 7.0~7.2 (m, 4H), 9.4 (broad, 1H).
5. 3a-ij was chromatographed on silica gel with 1:1 CHCl_3 and ethyl acetate (Tlc Rf=0.37). mp $205\sim 207^\circ$. IR (KBr disk) 3400, 3250, 1615 cm^{-1} ; NMR ($\text{DMSO}-d_6$ with TMS) δ 1.1~1.9 (m, 10H), 3.20 (broad, 1H), 3.7 (m, 1H), 6.8~7.2 (m, 3H), 7.78 (d, 1H), 7.9 (m, 1H), 11.1 (m, 1H).
3b-i was chromatographed on silica gel with ethyl acetate (Tlc Rf=0.56). mp 178° . IR (KBr disk) 3280, 3220, 1102 cm^{-1} ; NMR ($\text{DMSO}-d_6$ with TMS) δ 1.2~2.1 (m, 10H), 4.45 (m, 1H) 6.9~7.3 (m, 3H), 7.74 (d, 1H), 8.1 (m, 1H), 8.8 (broad, 1H), 11.2 (m, 1H).
6. 4a-i was isolated by distillation (bp $100^\circ/0.4$ mmHg). IR (neat) 1688, 1621 cm^{-1} ; NMR (CDCl_3 with TMS) δ 0.88 (t, 3H), 1.1~1.5 (m, 4H), 3.33 (s, 2H), 3.51 (t, 3H), 7.14 (s, 4H), 7.26 (s, 1H).
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